



Annual Review of Environment and Resources

Methane and Global Environmental Change

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Abstract

Global atmospheric methane concentrations have continued to rise in recent years, having already more than doubled since the Industrial Revolution. Further environmental change, especially climate change, in the twenty-first century has the potential to radically alter global methane fluxes. Importantly, changes in temperature, precipitation, and net primary production may induce positive climate feedback effects in dominant natural methane sources such as wetlands, soils, and aquatic ecosystems. Anthropogenic methane sources may also be impacted, with a risk of enhanced emissions from the energy, agriculture, and waste sectors. Here, we review the global sources of methane, the trends in fluxes by source and sector, and their possible evolution in response to future environmental change. We discuss ongoing uncertainties in flux estimation and projection, and highlight the great potential for multisector methane mitigation as part of wider global climate change policy.



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1. INTRODUCTION

Methane (CH_4) is a powerful greenhouse gas, with human activities, such as livestock production and fossil fuel extraction, having become an increasingly important source of emissions since the Industrial Revolution (1). Since 1750, atmospheric concentrations have more than doubled compared to the preindustrial average and, up until the late twentieth century, there was deepening concern about their growing role in global climate forcing (2). However, around the turn of the new millennium, atmospheric CH_4 concentrations appeared to stabilize, raising hopes that their sustained growth in previous decades had come to an end (3). In fact, the period of stable concentrations was short-lived, with a resumption of increasing atmospheric CH_4 from 2007 onward (4, 5). The cause of these swings in net annual fluxes remains an area of intense debate and research (6, 7). More certain is that CH_4 has become a key agent of global climate forcing and that further increases in its atmospheric concentration in the twenty-first century risk undermining international climate change mitigation efforts.

Here we examine CH_4 and global environmental change, our primary focus being on responses to and interactions with climate change and elevated atmospheric CO_2 concentrations. We first review methane's direct and indirect role in climate forcing, its global sources and sinks, and past trends in fluxes and atmospheric concentrations. We then examine the ways in which natural and anthropogenic CH_4 fluxes are affected by environmental change and the potential feedback responses to future climate change. Finally, we reflect on the role of CH_4 in global climate change policy and the multisectoral potential for its mitigation.

1.1. Direct Radiative Forcing

Since the Industrial Revolution, the direct radiative forcing of increased CH_4 is estimated to have been 0.48 Wm^{-2} (compared to approximately 1.8 Wm^{-2} for carbon dioxide) (8). Atmospheric

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Table 1 Evolving GWP estimates for CH₄ in the IPCC Assessment Reports FAR through AR5

Time horizon	FAR (1990)	SAR (1995)	TAR (2001)	AR4 (2007)	AR5 (2013)
20 years	63	56	62	72	84 (86) ^{a,b}
100 years	21	21	23	25	28 (34) ^b
500 years	9	6.5	7	7.6	NA

Abbreviations: AR4, Fourth Assessment Report; AR5, Fifth Assessment Report; FAR, First Assessment Report; GWP, global warming potential; IPCC, Intergovernmental Panel on Climate Change; NA, not applicable; SAR, Second Assessment Report; TAR, Third Assessment Report.

^aValues in parentheses include climate-carbon cycle feedback effects.

^bFor fossil fuel CH₄, these GWP values are increased by 1 and 2 for time horizons of 20 and 100 years, respectively (177), to account for the CO₂ released during combustion. However, where the combusted CH₄ is biogenic, or where the CO₂ from fossil CH₄ combustion is already accounted for, no adjustment is required.

CH₄ concentrations increased from 722 ppb to more than 1,800 ppb by 2011, largely as a result of human activities (9). That CH₄ has such a powerful direct radiative forcing effect despite its relatively low concentrations is due to its very effective absorption of infrared radiation, particularly at wavelengths of 3.3 and 7.7 μm . Importantly, CH₄ emissions also have major indirect effects on net radiative forcing, with these complex impacts making methane's overall role in global climate change greater still (10).

Methane's global warming potential (GWP; a measure of the warming induced by emission of any greenhouse gas relative to that of CO₂ on a mass basis) has seen substantial revision over the past 20 years. In 1990, the GWP for CH₄ was estimated as being 21 over a 100-year time horizon (11) (i.e., that each metric ton of CH₄ emitted had a warming effect equivalent to 21 metric tons of CO₂). This estimate has both increased and become more nuanced in subsequent Intergovernmental Panel on Climate Change (IPCC) Assessment Reports (9), as understanding of the indirect climate forcing impacts of CH₄ has improved and background concentrations have continued to change (**Table 1**). Current national emissions reporting under the United Nations Framework Convention on Climate Change utilizes the 100-year time horizon value of 25 for CH₄ from the Fourth Assessment Report.

1.2. Indirect Forcing

The primary sink for CH₄ in the atmosphere is oxidation by hydroxyl (OH) radicals (12). As CH₄ concentrations increase, this atmospheric oxidizing capacity becomes more exhausted, meaning the effective lifetime and radiative forcing of CH₄ is enhanced. In addition to this positive feedback effect, CH₄ has further indirect impacts through its role as a precursor for ozone formation and through the additional climate forcing that results from the water vapor and carbon dioxide produced when it is destroyed (13). The atmospheric lifetime of CH₄ is approximately 9 years, but with an effective perturbation lifetime of 12 years (due to the feedback effect CH₄ has on its own lifetime) (8).

In the most recent IPCC Assessment Report two different estimates for the GWP of CH₄ over a 100-year time horizon are given, with a value of 28 when climate-carbon feedbacks (such as the enhancement of the atmospheric lifetime of CO₂ as a result of warming induced by CH₄) are excluded, and a value of 34 when such complex feedbacks are included (8). Methane can also play an indirect role in global radiative forcing through its interactions with sulfate aerosols and the lifetimes of powerful greenhouse gases such as hydrofluorocarbons and hydrochlorofluorocarbons (14). The intersecting domino effects that are the atmospheric interactions of CH₄ mean that current uncertainties in GWP remain large (~40%).

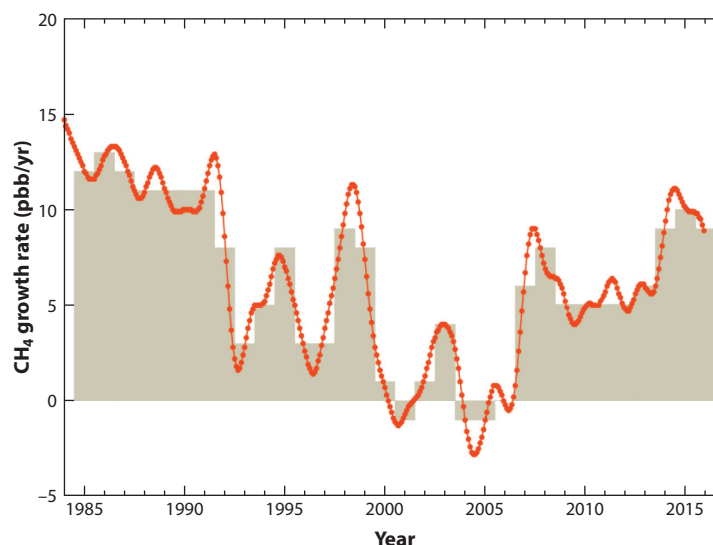


Figure 1

Globally averaged methane (CH_4) growth rate from 1984 to 2016. Shaded columns represent increases in successive annual means. Red dots represent monthly means (23). Figure reproduced with permission from the World Meteorological Office.

1.3. Past Trends and Current Concentrations

Through ice core records we are able to reconstruct global atmospheric CH_4 concentrations over the past 400,000 years and more (15). These indicate swings in concentration between approximately 350 and 750 ppb coinciding with glacial and interglacial periods, respectively. Over these long (20–100,000-year) time intervals, the cyclical changes in CH_4 concentrations can be broadly explained by Earth's orbit and consequent variations in solar input. However, within these orbital periods, shorter-term variations in CH_4 fluxes attributed to monsoon impacts on tropical wetlands can be observed (16). Recent analysis has also linked variations in tropical wetland CH_4 emissions with the El Niño–Southern Oscillation, with decreased CH_4 emissions associated with El Niño, and increased emissions with La Niña conditions (17). Such an interaction may help to explain lower growth rates in atmospheric CH_4 concentrations observed in the 1990s and early 2000s, as this period coincided with several El Niño events (18).

In addition to responses of wetland CH_4 fluxes to changes in temperature and precipitation, natural CH_4 sources such as permafrost soils, lakes, and coastal regions may have all played significant roles in determining the preindustrial CH_4 concentration (19). Indeed, loss of large amounts of CH_4 from marine clathrate deposits in response to warming (the clathrate gun hypothesis) has in the past been suggested as the primary cause of increases in atmospheric CH_4 concentrations during the Quaternary (20). Although the clathrate gun hypothesis remains controversial (21), a good understanding of how environmental change affects natural CH_4 sources is vital in terms of robustly projecting future fluxes under a changing climate.

In more recent history, and in particular since the nineteenth century, the atmospheric CH_4 concentration has risen far beyond the preindustrial average, with a growth rate reaching as high as 10–15 ppb year^{-1} in the late 1980s and early 1990s (22, 23) (**Figure 1**).

As mentioned above, a brief hiatus in atmospheric concentration increase around the turn of the millennium has since given way to renewed growth at 5–10 ppb year^{-1} . Although the

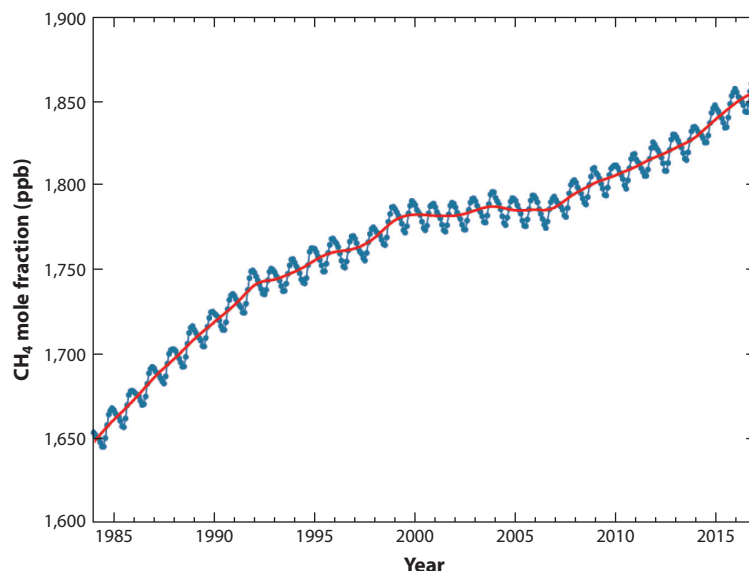


Figure 2

Globally averaged methane (CH_4) mole fraction in the atmosphere from 1984 to 2016. The red line is the monthly mean mole fraction with the seasonal variations removed; the blue dots and line depict the monthly averages. Observations from 125 stations have been used for this analysis (23). Figure reproduced with permission from the World Meteorological Office.

approximately 150% increase in CH_4 concentrations since 1750 is largely attributable to human-induced emissions, interannual variability in global fluxes and growth rates is also highly dependent on the balance of natural sources and sinks (4).

Recent measurements of atmospheric CH_4 show concentrations reached a new high of 1,853 (± 2) ppb in 2016. This represents a 9-ppb annual rise and a 157% increase in abundance relative to 1750 (23) (**Figure 2**).

The continued increase in CH_4 concentrations has been linked to enhanced wetland emissions (24). However, more recent estimates indicate their role in this increase is a minor one and that elevated emissions from agriculture and fossil fuel extraction may be more important (6, 25, 26). A decrease in OH concentrations—the primary sink for CH_4 —is also believed to have enhanced CH_4 in the atmosphere (7, 12). Importantly, several studies have highlighted the limitations of current surface observation networks (7, 27), inventory estimates (5), and models (28) in robustly attributing the sources of year-to-year variation in atmospheric CH_4 concentrations.

2. GLOBAL METHANE SOURCES AND SINKS

2.1. Methane Sinks

The atmosphere represents the dominant global sink for CH_4 , being responsible for the destruction of approximately $600 \text{ Tg CH}_4 \text{ year}^{-1}$ (**Figure 3**). Of this, reaction with OH in the troposphere is the main (85–90%) mechanism of atmospheric loss, with stratospheric OH destruction and reaction with tropospheric chlorine (Cl) making up the rest (22). Soils are also a significant global CH_4 sink, estimated at approximately 30 Tg year^{-1} . This net exchange figure masks the much greater levels of coupled CH_4 production and consumption that can occur within soil profiles (29).

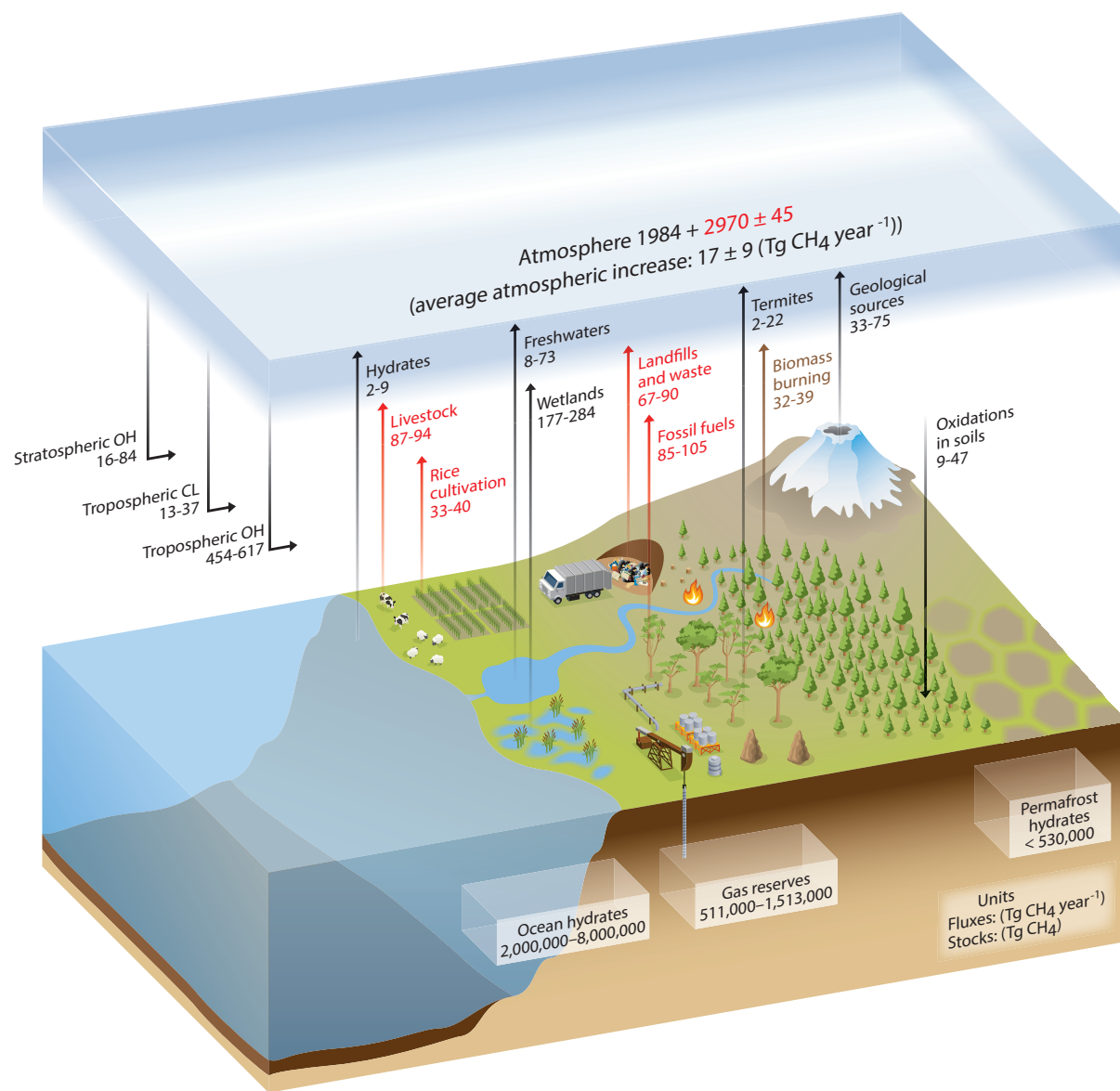


Figure 3

Global methane stocks and fluxes (22). Numbers represent annual fluxes in Tg CH₄ year⁻¹ estimated for the 2000–2009 time period and CH₄ reservoirs in Tg CH₄. Black arrows denote “natural” fluxes, that is, fluxes that are not directly caused by human activities since 1750, red arrows anthropogenic fluxes, and the light brown arrow a combined natural plus anthropogenic flux. Figure reproduced with permission from the Intergovernmental Panel on Climate Change.

Environmental change has the potential to affect all CH₄ sinks (30). For the atmosphere, changes in OH radical availability in the troposphere would have the greatest impact. Increases in UV radiation (due to stratospheric ozone depletion), humidity and temperature (due to climate change), and ozone and NO_x concentrations (due to both climate change and anthropogenic emissions) can all increase OH abundance (13). However, increasing CH₄ concentrations, together with

elevated carbon monoxide (CO) and volatile organic compound (VOC) emissions—themselves having significant anthropogenic sources (31, 32)—can decrease OH abundance. Since the Industrial Revolution, no clear trend in OH abundance at the global scale has emerged. However, for climate change impacts specifically, it is estimated that warming between 1850 and today has served to slightly increase OH abundance. This has effectively reduced the atmospheric lifetime of CH₄ by approximately four months and as such represents a weak negative feedback effect (33).

For the soil CH₄ sink, changes in land cover, temperature, precipitation, and deposition rates of nitrogen (N) and sulfur (S) may all alter local and regional sink strengths (34–36), with consequences for the magnitude of the global sink (see Section 3.2).

2.2. Methane Sources

Global sources of CH₄ are far more diverse than the sinks. Total emissions are currently of the order of 500–600 Tg year^{−1}, with anthropogenic sources now estimated to outweigh natural sources (Figure 3) (22).

2.2.1. Natural methane sources. Natural CH₄ sources include wetlands, geological sources, lakes and rivers, termites, wildfires and wild animals. Wetlands dominate these natural emissions, at between 177 and 284 Tg year^{−1}. As the largest CH₄ source type, interannual changes in wetland CH₄ fluxes are the primary driver of variations in global CH₄ flux (37). Wetlands encompass a host of ecosystems ranging from flooded forest soils and swamps in the tropics to high latitude peatlands and bogs. Methane emissions from these seasonally or permanently flooded soils can be very sensitive to environmental change, with CO₂ enrichment, temperature, and precipitation change being especially important (38). For example, the pause in growth of atmospheric CH₄ concentrations observed around the start of the new millennium has been attributed to a predominance of drier conditions over some northern and tropical wetlands at that time (17, 39).

The relative magnitude of wetland CH₄ sources at a global scale makes understanding their interactions with environmental change a vital component of emissions projections (40). However, their spatial distribution and diversity represent major challenges. Uncertainties around responses to changing plant cover and nutrient deposition, along with those for temperature and precipitation, mean that confidence in projected CH₄ emissions in the twenty-first century remains low (22) (Section 3.1).

Of the other natural CH₄ sources, all have the potential to grow or diminish in response to environmental change. Freshwater systems may see alterations in key determinants of CH₄ production, such as temperature, carbon, and nutrient inputs (41) (Section 4.2). Similarly, large geological CH₄ stocks such as marine hydrates may experience increasing temperatures and as such greater instability (42) (Section 4.1). For terrestrial ecosystems, changes in precipitation, temperature, and vegetation could directly and indirectly affect CH₄ fluxes from wildfires (Section 3.3) and in soils (Section 3.2).

A novel natural source of CH₄ has been previously reported for vegetation under aerobic conditions (43), with a suggestion that exposure of plant material to UV radiation may be an important mechanism in such aerobic methanogenesis (44). Changes in UV exposure—such as those that may occur with stratospheric ozone depletion and recovery—may therefore alter the strength of such a CH₄ source. However, more recent estimates suggest its global magnitude is small (0.2–1.0 Tg year^{−1}) (45).

2.2.2. Anthropogenic methane sources. Anthropogenic CH₄ sources are dominated by agriculture and energy-related emissions (approximately 130 and 100 Tg year^{−1}, respectively), with

waste responsible for a further 70–90 Tg year⁻¹ and biomass burning approximately 35 Tg year⁻¹ (**Figure 3**) (22). In the agriculture sector, ruminant livestock are the largest CH₄ source, with the potential for direct (e.g., animal welfare) and indirect (e.g., altered feed composition) impacts from future environmental change (Section 5.1). Rice agriculture—itsself having global CH₄ emissions of approximately 36 Tg year⁻¹—may be especially sensitive to environmental change, with variations in temperature, water availability, and CO₂ concentrations all having the potential to alter fluxes (Section 5.2).

A potentially important indirect impact of climate change on CH₄ emissions from agriculture is through increased loss and waste along the food supply chain. Severe weather and pest damage can reduce productivity, and higher temperatures and humidity may accelerate postharvest food deterioration (46). For commodities with relatively high CH₄ emissions intensities, such as some rice, beef, and dairy products, any substantial enhancement in such loss and waste could decrease overall production efficiency and drive up CH₄ emissions intensity still further (47).

Energy-related CH₄ emissions largely result from fugitive emissions during fossil fuel extraction and supply (48). Warming in permafrost areas has the potential to directly increase gas pipeline leakage rates, and ice retreat in the Arctic may also result in greater fossil fuel exploitation (including that of CH₄ hydrates) and as such further increases in fugitive CH₄ emissions (Section 5.3). Hydroelectricity generation can also result in substantial CH₄ emissions via river damming and CH₄ production in the resulting reservoirs. Elevated temperatures and changing carbon and nutrient inputs may then further alter these emissions. As with wildfire CH₄ emissions, the amounts of CH₄ arising from human-induced biomass burning can be altered due to changes in precipitation, temperature, and the composition of the biomass itself.

Finally, waste-related CH₄ emissions may be affected by changes in climate, with higher temperatures tending to enhance CH₄ production from waste decomposition. Under controlled conditions (e.g., anaerobic digestion) this may enhance CH₄ yields for energy use (49), whereas in more open systems (e.g., constructed wetlands, waste treatments, and landfill sites) there is the potential for an increase in CH₄ emissions to the atmosphere. For livestock manure in particular, increased temperatures may enhance CH₄ production (50) and, in open storage systems, enhance emissions (51) (Section 5.1).

3. GLOBAL CHANGE AND TERRESTRIAL METHANE FLUXES

3.1. Wetland Methane and Global Change

Wetland ecosystems are an important component of the global carbon cycle, as Northern wetlands (those at >45°N) store more than 50% of global soil organic carbon, due to the slow organic carbon decomposition rates that result from wet surface conditions and low temperatures (52). Wetlands are also the largest single source of atmospheric CH₄, with current annual emissions estimated at 140–280 Tg CH₄ year⁻¹ (53).

Key determinants of wetland CH₄ emissions include temperature, soil moisture, aerenchyma transport (the movement of CH₄ from soils to the atmosphere via the porous tissues in plant roots and stems), and substrate availability (28, 54). As such, any significant change in wetland CH₄ emissions in response to climatic change and CO₂ fertilization impacts would have major consequences for global emissions and atmospheric CH₄ concentrations.

For example, future climatic warming, particularly at high latitudes, could lead to a significant increase in net CH₄ emissions from wetlands, serving as a significant positive feedback to warming of the global climate system (55). Similarly, increased net primary production and soil water content could enhance methanogenesis (the microbial basis of wetland CH₄ production) and CH₄

emissions. However, more variable precipitation and drying of wetland soils—such as is likely to occur in the subtropics—could reduce wetland extent, limit methanogenesis, and as such reduce net emissions (56). One of the largest uncertainties in wetland CH₄ estimates is how wetland extent or inundated area will change over time (28, 53).

Estimating the magnitude of these responses at a global scale is extremely challenging given the limited spatial and temporal coverage of direct observations globally, the importance of local conditions and microbial communities in determining net fluxes, and the impacts of human activity of wetland extent and function (57). Land use activities such as grazing, drainage, and fertilization may all radically alter carbon cycling and net CH₄ fluxes. An assessment of wetland CH₄ emission in China over the past 30 years found that, whereas climate change enhanced wetland emissions (through increased temperature at high altitudes and increased precipitation in arid areas), human-induced reduction in wetland area led to an overall decrease in CH₄ emissions (58). Similarly, differences in land management history and internal characteristics of peatlands (such as open degraded peatlands compared to intact peatland forests) may alter the sensitivity of CH₄ emissions to temperature change (59).

To date, projections of wetland CH₄ flux at a global scale in the twenty-first century have tended to focus on potential climate change impacts (54), rather than attempting to integrate these with scenarios of land use change and human activities relevant to wetland CH₄ emissions at the local scale. The complexity involved in such an integrated assessment of wetland CH₄ at a global scale means that future human activity remains a large source of uncertainty (57).

Recent modeling has overtly included scenarios of climate change mitigation by examining the potential response of regional and global wetland CH₄ emissions under the four representative concentration pathways (RCPs) (60). Outputs indicate a substantial increase—from 172 ± 12 Tg CH₄ year⁻¹ currently to 221.6 ± 15 by 2100—even for the lowest RCP (RCP 2.6). Under the business-as-usual (BAU) scenario (RCP 8.5), total wetland area increases by 13% and global CH₄ emissions almost double relative to current levels.

For high latitude (boreal) wetlands, higher temperatures, winter thawing, and a consequent increase in soil moisture content were found to be the primary drivers of elevated emissions. Whereas for tropical wetlands, higher temperatures and changing precipitation patterns result in higher and more annually variable emissions, despite a slight decrease in tropical wetland extent (60).

The high temporal variability common to wetland CH₄ fluxes—something likely to increase with climate change—poses a particular challenge for their robust quantification and projection. Assessment of surface warming and moisture availability impacts on northern latitude CH₄ emissions using satellite-derived fractional inundation (61) highlighted the importance of fine (submonthly) temporal resolution of fractional inundation on annual CH₄ emission estimation.

3.2. Soil Methane and Global Change

Section 3.1 deals with wetland/peatland soils (e.g., histosols), whereas this section deals with mineral soils (all those that are not wetlands/peatlands). In mineral soils, there are micro-organisms that produce CH₄ (methanogens) and micro-organisms that consume (and oxidize) CH₄ (methanotrophs; see 56). The balance between methanogenesis and methanotrophy determines the net flux of CH₄ from soils (30, 56).

Aerobic mineral soils act as a sink for CH₄, oxidizing 9–47 Tg CH₄ year⁻¹ globally (22). Mineral soils under natural vegetation and forests tend to act as the strongest sink, followed by grasslands, with the sink strength weakest in cultivated soils and those receiving nitrogen fertilizer (62, 63); as such, as cropland has expanded, the CH₄ sink strength of soils globally will have declined (63). When mineral soils become anaerobic, the net flux to the atmosphere can be positive, with



waterlogged soils becoming a CH₄ source, often with large local emission rates (64). When soils are deliberately flooded, e.g., for rice cultivation, they can become very large global sources of CH₄ (see Section 5.2).

Permafrost soils contain large quantities of CH₄ in the form of hydrates, known as clathrates when trapped in permafrost, with a stock of the order of 500,000 Tg CH₄ (22), globally. Whereas permafrost soils are a relatively minor natural source of CH₄ emissions, estimated to contribute approximately 1 Tg CH₄ year⁻¹ (22), permafrost thaw, driven by climate change, could release a portion of this vast CH₄ store to the atmosphere (65, 66).

In the tropics and subtropics, termites are a significant natural source of CH₄ emissions. Where termites occur, CH₄ oxidizing soil organisms are also present, meaning that a large proportion of the CH₄ produced by termites is oxidizing in mound material (67). Not all of the CH₄ is oxidized, however, meaning that globally termites are a net CH₄ source totaling 2–22 Tg CH₄ year⁻¹ (22).

In terms of latitudinal differences in CH₄ fluxes, Arctic soils tend to act as a CH₄ source (65), as do termite-rich tropical/subtropical regions (22), whereas mineral soils in temperate regions tend to act as a CH₄ sink (unless waterlogged; see 63), with 30–50% of the global soil CH₄ sink located in the temperate zone (68). The positive north–south gradient in CH₄ concentrations, however, is mostly driven by the greater emissions of CH₄ from wetlands and anthropogenic emissions in the northern compared to the southern hemisphere (22).

Soil-mediated fluxes of CH₄ can be very sensitive to climate change and variability (22). Dlugokencky et al. (24), for example, showed that the high CH₄ growth rate during 2007 to 2008 was associated with positive precipitation and temperature anomalies. Warming would likely increase soil CH₄ fluxes in the Arctic through mobilization of CH₄ hydrates resulting from permafrost thawing (65), although, as with wetlands (Section 3.1), it might decrease emissions from otherwise waterlogged mineral soils in some regions if it dried the soils through increased evapotranspiration. Changing precipitation patterns will have mixed effects, with drying regions expected to emit less CH₄ and wetting regions (if already wet) expected to emit more CH₄ due to greater incidence of waterlogging. According to the IPCC (22), soil CH₄ oxidation is projected to increase overall (by up to 23% under a moderate climate forcing scenario), attributable to rising atmospheric CH₄ concentrations, increased soil temperature, and decreased soil moisture (69, 70).

Land management affects the sink strength in temperate mineral soils, with more intensive activity (i.e., cultivation for crops) decreasing the sink strength to the greatest degree (63; also see Section 5.2, where this is discussed further). Although wetland CH₄ emissions have been observed to increase in response to elevated atmospheric CO₂ concentrations, this has been attributed to increasing soil moisture due to the reduced plant demand for water under higher CO₂ (71), and emissions of CH₄ from permafrost are expected to increase due to CO₂ fertilization (72), the magnitude of any such effect on mineral soils is not clear. Nitrogen loading is known to decrease the CH₄ oxidation potential of soils (73), but the impacts of changes in nitrogen deposition, as with CO₂ fertilization, are complex and remain highly uncertain (63).

3.3. Wildfire Methane and Global Change

Wildfire CH₄ emissions show high interannual variability, with a global estimate of 3 Tg CH₄ year⁻¹ and a range of 1–5 Tg year⁻¹ (22). Climate change, land use, and fire management practices all have the potential to radically affect such emissions; CH₄ emissions arise from incomplete combustion of biomass and changes in vegetation, fire intensity, timing, and distribution, which are key determinants of this. Climate change projections suggest an increase in both wildfire frequency and intensity in regions such as North and South America, central Asia, and southern Africa (74). The projected increase is primarily driven by higher temperatures, with changing precipitation also being an important determinant in the subtropics (75).

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Table 2 “Likely” response of CH₄ fluxes from major natural sources in response to global change in the twenty-first century^{a,b}

Source	Temperature response	Precipitation response	CO ₂ response	Overall
Soils and wetlands (low latitude)	↗	→	↗	↗
Soils and wetlands (high latitude; >45°)	↗	→	↗	↗
Wildfires	→	→	↗	→
Marine	↗	→	→	↗
Freshwater	↗	→	↗	↗

^aThese are “likely” responses in the view of the authors, assuming no confounding impacts from other human activities such as wetland drainage.

^bResponses represent global averages unless otherwise specified.

A global increase in wildfire area, all else being equal, can be expected to increase CH₄ emissions from this source, although the emissions intensity of wildfires is highly dependent on factors such as biomass type and moisture content (76). Given that climate, land use change, and CO₂ enrichment will also alter vegetation characteristics in many wildfire-prone regions, it is possible that the CH₄ emissions intensity of wildfires will increase in some areas (such as tropical peatlands and forests) (77), while decreasing in others (e.g., subtropical savannas). However, at a global scale recent estimates suggest wildfire emissions (in the form of CO₂) declined in the twentieth century and are unlikely to greatly increase in the twenty-first century due to climate change except under a high emissions scenario (i.e., RCP 8.5) (78) (Table 2). Finally, it has been suggested that wildfires can alter the magnitude of the soil CH₄ sink through driving vegetation change, reducing soil organic matter content, and altering gas diffusion rates (79, 80).

4. GLOBAL CHANGE AND AQUATIC METHANE FLUXES

4.1. Oceanic Methane Fluxes and Global Change

Compared to other natural and anthropogenic emissions, the oceans are a modest source of CH₄ to the atmosphere. Globally, they contribute approximately 30 Tg CH₄ year⁻¹ (81); this estimate includes CH₄ emissions from the open ocean (~2 Tg CH₄ year⁻¹), shelf seas (~6 Tg CH₄ year⁻¹), and estuaries (~2 Tg CH₄ year⁻¹), as well as inputs from seafloor seeps (hydrocarbon seeps, mud volcanoes, hydrothermal vents) (~20 Tg CH₄ year⁻¹) (81). Although shallow marine sediments host vast quantities of CH₄ (>500 Pg C as CH₄; see 82) in the form of marine clathrate, these are not considered to be a significant source of atmospheric CH₄ at present (83). The uncertainty in all of these estimates is, however, high due to the limited number of measurements, the ephemeral nature of seafloor seeps, and difficulties in quantifying ebullition fluxes of CH₄.

Methane is produced in ocean sediments by thermal and anaerobic microbial degradation of organic carbon, as well as abiotically during high-temperature serpentinization of ultramafic rocks in the lower oceanic crust by circulating seawater. Methane is also produced within the ocean in oxygen and sulfate-replete surface waters. This so-called marine methane paradox has until recently been attributed to methanogenesis within anoxic microenvironments (such as digestive tracts and freshly released fecal pellets; see 84), but it is now clear that aerobic bacterial degradation of phosphonate constituents of dissolved organic matter can produce enough CH₄ to support the entire sea-air CH₄ flux at least at some locations (85). Aerobic production of CH₄ from CH₃SH by bacteria is also thermodynamically plausible (86).

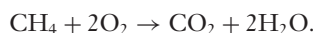
Anthropogenic sources of CH₄ are growing in importance. Most of the CH₄ in the Yaquina estuary (USA) originates from the release of municipal wastewater (87), and CH₄ emissions from Guanabara Bay (Brazil) are dominated by sewage discharge (88). Fugitive emissions from oil and

gas extraction are also a concern: In the North Sea, CH₄ is still being released from an abandoned drill site more than 20 years after a major blowout (89), and the blowout of the Deepwater Horizon oil well in the Gulf of Mexico in April 2010 injected up to 500,000 metric tons of natural gas, mainly CH₄, into the deep sea (90).

Emissions of CH₄ from the oceans to the atmosphere are strongly moderated by aerobic and anaerobic microbial oxidation (91). Anaerobic oxidation of methane (AOM) occurs in anoxic seawater and sediment pore waters, and is believed to involve a consortia of archaea and sulfate-reducing bacteria:



After reduction by photochemical processes in the troposphere, AOM is the largest sink of CH₄ on our planet (92). However, if sedimentary CH₄ fluxes are high then the rate of sulfate diffusion into sediments limits the rate of AOM, and/or CH₄ solubility in the pore waters may be exceeded, such that free CH₄ gas is formed and CH₄ escapes into the overlying water column. Methane transported into oxic sediment pore waters and seawater, and CH₄ produced within the oxic water column, may be oxidized aerobically:



Methane oxidation rates are widely variable, with turnover times of the order of days to >1,000 years (93), depending on the availability of dissolved CH₄ and O₂, as well as temperature, salinity, and hydrographic dynamics. Most CH₄ bubbles emitted from sediments at depths >200 m do not reach the surface mixed layer of the ocean (where CH₄ can be exchanged with the atmosphere) because of bubble dissolution and CH₄ oxidation (94). However, CH₄ bubbles from hydrocarbon seeps can be coated with a thin layer of oil that inhibits dissolution and enables them to persist to much greater heights above the seafloor (550–600 m) (95).

The majority of the global CH₄ flux from the oceans comes from the continental shelf, coastal seas, and estuaries (96, 97) due to high rates of methanogenesis sustained by high biological productivity and organic matter sedimentation, as well as direct inputs of CH₄ from rivers. Projecting future emissions from these areas is difficult, given uncertainties in current estimates and the multiplicity of processes that moderate CH₄ emissions. Increased temperatures will stimulate microbial CH₄ production and decrease CH₄ solubility, increasing CH₄ emissions from the seabed. Higher seawater CH₄ concentrations, however, mean that CH₄ oxidation rates will also be higher (93). Nevertheless, it is likely that increased emissions from shallow gassy sediments beneath a well-mixed water column will increase CH₄ fluxes to the atmosphere and provide a positive feedback on climate (98). If mangroves are confirmed to be a significant source of atmospheric CH₄ (99), then their continued removal may, in part, mitigate increased coastal CH₄ emissions.

Although coastal eutrophication may both increase CH₄ and lower O₂ (100), oxygen-deficient zones (both in the coastal and open ocean) do not appear to be a significant source of atmospheric CH₄ at present, nor are they predicted to be in the future, even though these zones are expected to expand as climate warms (101).

The effects of global change on CH₄ production and consumption in the open ocean are essentially unknown. There is evidence that the Atlantic overturning circulation is slowing (102), and the upper ocean becomes more stratified, at most locations, in response to global warming (103), which may mean that a higher proportion of the flux from sediments is oxidized below the thermocline. Increased sea surface temperatures may enhance primary productivity and in turn CH₄ production in anoxic microenvironments, although stratification may reduce return of remineralized nutrients to surface waters (limiting primary production) and higher temperatures will increase the rate of aerobic CH₄ oxidation. Nevertheless, it is now clear that CH₄ production in aerobic surface

waters is directly linked to nitrogen, phosphorous, and carbon cycles (86), thus any change in the operation of these biogeochemical cycles will have a profound effect on CH₄ emissions.

Sea ice acts as a physical barrier to CH₄ exchange with the atmosphere (104), and concentrations of atmospheric CH₄ have been shown to increase over open leads and regions with fractional sea ice cover (105). High latitudes of the northern hemisphere are expected to experience a larger temperature increase than other regions due to climate change (9), and sea ice coverage in the Arctic has decreased in recent decades, especially in the summer (106). The effect of this on sea-air CH₄ fluxes is manifold; the presence of sea ice reduces turbulence and wave generation in adjacent ice-free waters, restricting gas exchange (107). Melting may act to both increase gas exchange - as larger waves will be more likely and swells will be more common (108), and increase stratification between surface and deep waters. Such stratification would restrict the capacity of the deeper Arctic Ocean to act as a CH₄ sink (109). Although there appears to be a direct relationship between sea-ice decline and increasing CH₄ emissions in the Arctic, the contribution of oceanic CH₄ sources is still unclear (110).

Most modeling studies agree that warming of bottom waters on the continental shelf and slope will destabilize marine clathrate and consequently increase CH₄ fluxes across the seabed (111). In 2008, more than 250 plumes of CH₄ bubbles were discovered seeping from the seafloor offshore of western Svalbard, and these emissions have been attributed, at least in part, to clathrate dissociation as a result of warming of bottom waters in this area over the past ~30 years (112). Partial thawing of submerged permafrost in Arctic shelf seas has also been observed and appears to be a significant source of CH₄ to the atmosphere (113), although this is likely related to warming initiated by permafrost submergence approximately 8,000 years ago rather than recent Arctic warming (114). Release of large quantities of CH₄ into marine sediments is of major concern because the efficacy of both AOM and aerobic CH₄ oxidation is reduced as CH₄ will be transported primarily in the gas phase, which is largely inaccessible to microbes (115). Nevertheless, it seems unlikely that catastrophic, widespread dissociation of marine clathrates will be triggered by continued climate warming at contemporary rates (0.2°C per decade; see 9) during the twenty-first century (116).

Although knowledge of the oceanic distribution, formation and consumption, and sea-air fluxes of CH₄ has improved considerably in recent decades, their response to environmental change remains uncertain. Coastal regions are especially heterogeneous both temporally and spatially, and changes to freshwater inputs to the coastal zone are highly uncertain. Moreover, there is a paucity of studies on CH₄ distributions in the southern hemisphere (117), and it is predicted that thousands of natural CH₄ seeps remain to be discovered on the continental margins (118) that are likely to be affected by warming of overlying waters. Improved knowledge of the response of microbial communities to increased temperature and changes in CH₄ fluxes is critical—AOM communities, especially, grow very slowly and may take decades to establish (119).

4.2. Freshwater Methane and Global Change

Globally, CH₄ emissions from freshwater systems have been estimated to be as much as 100 Tg year⁻¹ (120). This estimate includes lakes, rivers, and reservoirs, with the latter representing an anthropogenic source of approximately 20 Tg year⁻¹ (see also Section 5.3). Uncertainty in such global estimates remains high due to the limited number of direct measurements and the ephemeral nature of some freshwater systems, such as river flood plains.

Methane production in freshwater systems is predominantly via microbial methanogenesis in sediments, although in highly turbid, low-oxygen systems CH₄ may also be produced in the water column (121, 122). Methane is then primarily emitted to the atmosphere via diffusion, ebullition, and, where emergent vegetation is present, through plant aerenchyma. In most systems, microbially mediated CH₄ oxidation in surface waters plays a key role in buffering net emissions



to the atmosphere. However high rates of emission are possible in areas where bottom waters supersaturated with CH₄ are exposed directly to the atmosphere, such as in hydroelectric turbines and spillways (120, 123).

Environmental change may alter CH₄ emissions from freshwater systems in numerous ways. Temperature is a key determinant of methanogenesis and thus, all else being equal, climate change would be expected to enhance sediment CH₄ production and emission. This positive feedback effect is of particular importance in high latitude lakes, with warming of Arctic lakes under a high emissions scenario having the potential to more than double CH₄ emissions in the twenty-first century (from approximately 12 Tg year⁻¹ today to more than 28 Tg year⁻¹ in the 2090s) (124).

Much of this enhancement derives from the thawing of carbon-rich permafrost at the margins of Arctic lakes, and a consequent fueling of methanogenesis in the lake sediments (41). However, this same thawing of permafrost is expected to lead to drainage of some lake and pond areas (especially in the far north) and expansion of others. As such, an overall reduction in the area of Arctic lakes may serve to limit the extent of this positive climate change feedback. Increasing temperatures can also result in increased stratification of freshwater systems, reducing oxygen availability in sediments and bottom waters (125) and serving to enhance methanogenesis in some systems.

Changes in precipitation can have both direct and indirect effects on CH₄ emissions from freshwater systems. Reduced rain and meltwater inputs may extend water residence times, enhance eutrophication and anoxia, and as such increase methanogenesis. However, reduced precipitation due to climate change may also lead to area reduction or complete loss of some freshwater systems (41) (most notably in the subtropics), resulting in an overall decrease in CH₄ emissions. Increases in precipitation, however, can greatly increase the area of freshwater systems. Where these waters inundate areas with organic soils or above-ground biomass—such as occurs seasonally in the Amazon basin (126) or in the creation of large reservoirs (127)—significant enhancements in CH₄ production and emission may occur.

An increase in the intensity of precipitation events is also likely to increase the amounts of allochthonous organic matter and macronutrients that many freshwater systems receive via leaching and runoff (41). Such additional inputs may then serve to fuel eutrophication, methanogenesis, and net increases in CH₄ emissions. Similarly, CO₂ enrichment and changes in plant biomass across catchments may serve to enhance carbon inputs to drainage waters and the lakes and rivers they feed. A trend of increasing dissolved organic carbon (DOC) inputs to freshwater systems in North America and Northern Europe has already been reported, with a prediction that DOC production in some catchments could rise by 20% in response to doubling of CO₂ (128).

The sheer diversity of freshwater systems around the world inevitably makes any kind of robust estimation of CH₄ fluxes difficult. There remains a paucity of direct CH₄ measurements, especially those that encompass ebullition and plant-mediated fluxes alongside diffusion losses (120). As such, quantitative projections of future CH₄ emissions from this source are even more challenging. Uncertainties around the impacts of environmental change, and in particular climate change, on freshwater CH₄ fluxes in the twenty-first century, are compounded by changing land use and catchment management practices (129). The only certainty is that CH₄ emissions from some freshwater systems will increase, whereas in others they will decrease, with the overall climate change interaction likely to be a weak positive feedback effect at a global scale (Table 2).

5. GLOBAL CHANGE AND ANTHROPOGENIC METHANE

5.1. Livestock Methane Emissions and Drivers

Methane emissions arise from livestock in two principal ways; enteric CH₄ arising as a by-product of the breakdown of ingested feed and CH₄ from animal wastes arising from the anaerobic

fermentation of stored and pasture-deposited animal urine and feces. Enteric CH₄ arises mainly from ruminant livestock (cattle, buffalo, sheep, and goats), whereas CH₄ emissions from animal wastes arise from all types of livestock, although dairy cattle and pigs are the most important.

Enteric CH₄ is produced under anaerobic conditions by a diverse community of methanogenic archaea using mainly hydrogen and CO₂ as substrates, although smaller amounts are produced using formate and methyl compounds as alternatives to hydrogen (130). The biggest single determinant of how much CH₄ an individual ruminant animal produces is the quantity of feed it eats, although the type and chemical characteristics of ingested feed also have an influence (131, 132). The animal itself can also influence the quantity of CH₄ it produces via host effects on digestive physiology and the makeup of the resident microbial population (130, 132–135).

Methane production from animal wastes is also an anaerobic microbial process and occurs mostly when animal wastes are stored. Lesser amounts are produced from wastes deposited directly onto the ground. Manure type (e.g., wet versus dry), storage method, storage duration, manure chemical composition, and temperature all influence the quantity of methane produced per unit of substrate.

Estimates of the quantity of CH₄ currently produced globally by livestock vary with the method of estimation and are currently in the range 90–120 Tg CH₄ year⁻¹ (22, 136–138). This is 6–7% of total global anthropogenic CO₂-e emissions and ~35% of total anthropogenic CH₄ emissions. Enteric CH₄ emissions comprise approximately 90% of all livestock-derived CH₄ emissions, with cattle (77%) being the dominant source (136). Manure management emissions are dominated by pigs (~42%) and cattle (~41%) (136).

Livestock CH₄ emissions have risen by just over 50% in the past 50 years with the largest increase occurring in Africa, Asia, and the Americas. Europe is the only region where emissions have decreased. An analysis by latitude of changes between 2000 and 2013 (137) suggests that the largest increases have occurred between latitudes 30°N and 30°S driven by increases in Central America, East Africa, and Brazil.

Forecasting future livestock emissions is highly uncertain as they will be influenced by the demand for livestock products. This in turn is driven by population growth, dietary preferences, the successful implementation of mitigation practices, and to a minor extent climate change itself, although studies exploring climate change impacts on livestock and their emissions remain limited (139–141). There is currently a strong consensus that CH₄ emissions will continue to increase under BAU scenarios (136, 139, 142, 143). The United Nations Food and Agriculture Organization estimates are that, by 2030, global livestock CH₄ emissions will be 12% higher than current emissions with the largest increases occurring in the Asian region (138). Europe is the only region where emissions are not forecasted to rise.

Given that there is considerable variability in emissions per unit of product within and between systems of production, there remains considerable scope for reductions in emissions intensity into the future and for increased efficiency of production to be an economically viable method of constraining absolute emissions below BAU scenarios (136, 144).

The high-level ways in which climate change could have an impact on future livestock CH₄ emissions have been well described (145). However, few studies have attempted to quantify how the impacts of, and adaptation to, changes in temperature, rainfall, and CO₂ concentration could modify emissions projections that are based primarily on population growth and increased per capita consumption of milk and meat.

With respect to enteric fermentation, climate change will not impact the fermentation process itself, given that internal body temperatures are well regulated. Elevated temperatures will, however, influence CH₄ emissions from livestock wastes as ambient temperatures do affect the

quantity of CH₄ produced; the temperature effect changes little once annual average temperatures reach 20°C (146), meaning that the largest impacts will be at higher latitudes.

The principle effects of climate change on livestock CH₄ emissions will be mediated through changes in the quantity and quality of feed available, heat stress, and the incidence of pests and diseases (147). These effects will work at the individual and herd/flock level.

With regard to feed supply, the impact on both crops and grasslands needs to be considered. In aggregate, the impact of climate change on grassland and crop production is generally small once the CO₂ fertilization effect is considered, although this hides large regional and local variation and uncertainties (147, 148). Grassland productivity may in fact increase under climate change, particularly at higher latitudes (148, 149) and altitudes (147).

Close to 50% of the feed consumed by livestock is grass, and the nutritive value of grass is influenced by temperature (150). Lower nutritive value (e.g., high fiber content) tends to increase CH₄ production per unit of feed eaten (131, 151), and more feed needs to be consumed to reach a given level of production. Taken together, these effects will increase emissions from both enteric and waste sources. The combined effect has not been globally quantified, although the impact of a reduction in herbage quality at a constant intake has been estimated to be an increase in CH₄ emissions of 0.9% for every 1°C increase in temperature (150).

Livestock are widely distributed globally and are therefore well adapted climatically. However, they are directly influenced by temperature. Higher temperatures increase the quantity of energy required to meet a given level of production and simultaneously reduce feed intake (152). Reproductive performance is also adversely affected by heat stress (153). Increased mortality in livestock has also been linked to heat stress (154). Changes in temperature and rainfall may also influence the type and severity of pests and diseases that affect livestock. Collectively, these effects are likely to reduce the productivity of livestock systems and thus result in increasing emissions intensity. The potential net effect of temperature change on future CH₄ emissions has not so far been quantified in any systematic way. Reductions in forage quality, heat-induced increases in energy demand, impaired reproductive function, and increased fermentation rates in stored animal wastes all have the potential to increase emissions. However, given the adaptability of livestock production systems, these drivers of CH₄ emissions are likely to be small compared with demand-induced increases in animal numbers, individual animal productivity, and potential constraints due to competition for land.

5.2. Cropland Methane and Global Change

As described in Section 3.2, aerobic mineral cropland soils have a much lower CH₄ sink strength than undisturbed or forest soils (63). The lower measured oxidation rates are attributed to soil disturbance and nitrogen fertilizer use (155), which suppress the soil methanotroph community (56). Historical land use change and a move to more intensive cropping would therefore be expected to have decreased the CH₄ sink strength of soils.

When soils are deliberately flooded for rice cultivation, they can become very large global sources of CH₄, with rice paddy emissions accounting for 33–40 Tg CH₄ year⁻¹ (22), a major anthropogenic emission source. The global cropland CH₄ emission profile of croplands is dominated by rice production, with 90% of emissions from tropical Asia, >50% from just China and India (156), and a small contribution to the global CH₄ soil sink from other croplands. Changes in rice management have the potential to significantly decrease paddy rice soil CH₄ emissions, with mid-season drainage now becoming prevalent in many rice-growing areas (144, 157). Other mitigation measures include changed fertilizer practices and tillage/residue management (144; see also Section 6).

As for mineral soils under natural vegetation, warming might decrease emissions from otherwise waterlogged mineral soils in temperate regions if it dried the soils through increased evapotranspiration, although croplands are less likely to become waterlogged than pastures. Given that global crop productivity is expected to be adversely affected by climate change at the global level (158), carbon returns to the soil could also decline thereby providing less substrate for methanogens, but the overall effect is not known. Ciais et al. (22) report that soil CH₄ oxidation is projected to increase under climate change due to rising atmospheric CH₄ concentrations, increased soil temperature and decreased soil moisture (69, 70).

Given that croplands are usually fertilized with nitrogen, one would expect changes in nitrogen deposition to have smaller effects on croplands compared to soils under natural vegetation, where N deposition will form the largest N input to the system. Although CO₂ fertilization might increase cropland productivity to some extent, at the global level, cropland productivity is projected to decline in the future under the combined impacts of climate change and increased atmospheric CO₂ concentration (i.e., climate change impacts will outweigh those of CO₂ fertilization). The impacts on future CH₄ fluxes arising from these complex interactions in croplands are not known.

5.3. Energy-Related Methane and Global Change

Energy-related CH₄ emissions are currently dominated by fugitive emissions during coal, oil, and gas extraction (159). The increase in shale gas exploitation over the past decade has added to such fugitive emissions, with high loss rates reported for some operations in the United States (160). Downstream CH₄ emissions can occur during processing (e.g., coal pulverization) and via leaks in the gas transmission network (161). The latter may be particularly sensitive to climate change due to thaw of permafrost soils and consequent fracturing of gas pipelines at high latitudes (162).

More widely, the retreat of sea and land ice in the Arctic as a result of warming is likely to result in increased exploitation of fossil fuel resources across the region (see, e.g., 163, 164). Such exploitation will inevitably lead to additional fugitive CH₄ emissions during extraction and transmission. A fast-emerging issue is the potential for large-scale exploitation of marine methane clathrates (CH₄ molecules enclosed in a frozen lattice of water molecules) found in continental shelf sediments. It is estimated that such clathrates constitute approximately 50% of all hydrocarbon resources, with an estimated 500–2,500 Pg C in the form of marine methane clathrates globally (82).

To date, extraction of these large CH₄ stocks has been uneconomic, but technological advances mean that commercial exploitation is now a real possibility (165–167). As discussed previously (Section 4.1), the stability of CH₄ clathrate deposits may already be at risk from climate change. Accidental or deliberate disturbance, due to fossil fuel extraction, has the potential for extremely high fugitive CH₄ losses to the atmosphere (168).

For hydroelectric power, reservoir creation and the inundation of organic soils and above-ground biomass is an established source of anthropogenic CH₄ emissions (41). The expansion of large-scale hydroelectric reservoirs may therefore increase such CH₄ emissions. Globally, hydroelectric reservoirs are estimated to emit approximately 3 Tg C as CH₄ each year, with the highest emissions occurring in tropical regions such as the Amazon basin (169).

As with natural freshwater systems, changes in precipitation, temperature, and plant growth (e.g., due to the CO₂ fertilization effect) can each affect CH₄ emissions from hydroelectric reservoirs (170). Projecting future emissions from these reservoirs is difficult, given uncertainties in current estimates and the impacts that catchment land use and climate change may have. Nevertheless, increasing temperatures are likely to enhance CH₄ production and emissions at many sites. More intense rainfall and accelerated primary production may also combine to increase allochthonous C and nutrient inputs, and thereby further fuel reservoir methanogenesis.



Table 3 “Likely” response of CH₄ fluxes from major anthropogenic sources in response to global change in the twenty-first century^a

Source	Temperature response	Precipitation response	CO ₂ response	Overall
Agriculture (direct) ^b	➔	➔	➔	➔
Agriculture (indirect) ^c	↗	↗	➔	↗
Energy (fugitive) ^d	↗	➔	➔	↗
Energy (hydroelectric)	↗	➔	↗	↗

^aResponses represent global average response unless otherwise specified.

^bDirect agriculture here refers to direct impacts of global change on CH₄ emissions from livestock, crop production, and agricultural wastes.

^cIndirect agriculture here refers to global change impacts on loss and waste in the food supply chain that result in upstream changes in agricultural CH₄ emissions.

^dFugitive sources here include existing fossil fuel extraction and transport, as well as potential disturbance and exploitation of methane clathrate deposits.

Current estimates of anthropogenic CH₄ emissions from energy-related biomass (including some nonenergy biomass burning) are approximately 35 Tg CH₄ year⁻¹ (22). The wide range of biomass types used for energy—from solid hardwoods and softwood pellets, to charcoal and peat, to rice straw and manure—together with a multitude of combustion methods, means such estimates are highly uncertain. As with wildfire CH₄ emissions, changes in biomass characteristics, such as moisture content, in response to environmental change may alter CH₄ emissions intensity. However, such changes in fuel quality are likely to be compensated for by improvements in combustion technology (e.g., clean cook stoves, cleaner-burn biomass-to-energy plants) (171). As such, this CH₄ source is likely to decrease in the twenty-first century even if total biomass volume burned increases (Table 3).

6. METHANE IN CLIMATE CHANGE MITIGATION

As underlying drivers of CH₄ emissions, such as energy and food demand, intensify in the twenty-first century, BAU anthropogenic CH₄ emissions are likely to rise further (48). Between 2010 and 2030, emissions from the energy sector are projected to increase by 25%, to approximately 130 Tg CH₄ year⁻¹. Similarly, a >15% increase in CH₄ emissions from the agriculture sector is expected under a BAU scenario by 2030, with expanding livestock production being the primary cause.

A sustained push toward low-CH₄ strategies in these key sectors is therefore vital if global CO₂ abatement policies aimed at avoiding dangerous climate change are not to be undermined. The role to arrest growth and then make deep cuts in global CH₄ emissions is substantial (Figure 4).

For the energy sector, there is significant potential for CH₄ abatement both in coal mining and in the oil and gas sector. For coal, strategies such as premine draining, postmine boreholes, and ventilation air methane (VAM) oxidation could more than halve CH₄ emissions relative to BAU in 2030 (reduction of 22 Tg CH₄ year⁻¹) at a cost of <\$100 per metric ton CO₂-e. In the oil and gas sector, reductions in approaching 50% compared to BAU in 2020 are also possible. Reduced emissions completions for shale gas, equipment maintenance and upgrading, and improved inspection regimes for leaks, could help to yield an overall cut in oil and gas sector emissions of approximately 50 Tg CH₄ year⁻¹ by 2030.

For CH₄ arising from waste, the greatest potential for mitigation at a global scale is centered on landfill. Here, a combination of landfill gas collection for energy generation or flaring, and greater diversion of waste to recycling and reuse, could reduce CH₄ emissions by more than 40% (a reduction of 19 Tg CH₄ year⁻¹ relative to BAU in 2030) at a cost of <\$100 per metric ton CO₂-e.

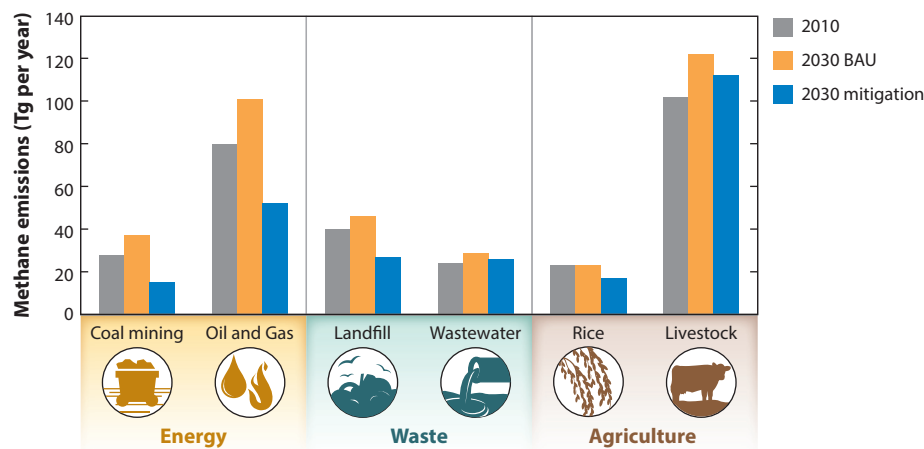


Figure 4

Global methane emissions and potential for mitigation across key sectors from 2010 to 2030. (48). Estimated emissions are in Tg CH₄ year⁻¹. Data are derived from estimates of CO₂-e by dividing by a GWP of 21. Red bars denote BAU emissions. Green bars denote emissions after all mitigation strategies with an abatement cost of <\$100 per metric ton CO₂-e are implemented. These include drainage of coal mine CH₄, landfill CH₄ collection, improved livestock health and diet, and manure management. Projected CH₄ emissions for agriculture have been disaggregated from other non-CO₂ emissions based on an assumption that CH₄ comprises 60% of non-CO₂ emissions from manure management and 65% of non-CO₂ emissions from rice cultivation (48). Abbreviations: BAU, business-as-usual; GWP, global warming potential.

Finally, for the agriculture sector, strategies addressing enteric fermentation and manure management emissions in livestock production could deliver cuts of >10 Tg CH₄ year⁻¹, and improved water and harvest residue management also have the potential for significant reductions in global CH₄ emissions from rice cultivation. Food supply chain and demand-side interventions that induce production-phase CH₄ mitigation, such as reduced supply chain loss and waste, also have an important role to play in this sector (172).

Overall, emission of >100 Tg CH₄ year⁻¹ could be avoided via such CH₄ abatement strategies—equivalent to a 30% cut by 2030 relative to a BAU (**Figure 4**). This is at an estimated cost of <\$100 per metric ton of CO₂-e. If maximum technical abatement potential is included this figure rises to near 40% (48).

The greenhouse gas emission reductions proposed by all nations under the Paris Climate Change Agreement (termed Nationally Determined Contributions or NDCs) encompass both long-lived greenhouse gases, such as CO₂, and short-lived climate pollutants such as CH₄. However, identifying specific CH₄ mitigation strategies within such NDCs is difficult, as planned emission reductions are usually referred to in terms of CO₂ equivalents (173). For some strategies, such as drainage management in rice agriculture, CH₄ mitigation is the tacit focus, whereas for strategies such as more efficient biomass burning a mix of long- and short-lived climate pollutants is combined.

The importance of addressing CH₄ and other short-lived climate pollutants as part of wider climate change mitigation efforts has been debated for many years (174–176). Substantial cuts in CH₄ emissions can have a relatively quick impact on the extent of warming compared to CO₂ reductions. However, over longer timescales (i.e., the late twenty-first century and beyond) reductions in CO₂ have a much greater effect on global temperatures.

In fact, aggressive and sustained reductions in both long- and short-lived climate pollutants are now required. The goal of the Paris Agreement is to limit global temperature rise in the twenty-first century to well below 2°C, while simultaneously promoting sustainable development and poverty reduction (173). Regarding the former, sustained CH₄ mitigation would complement CO₂ reductions and help bring the “ambition target” of 1.5°C within reach. Regarding the latter, rapid, large-scale reductions in CH₄ emissions could help ensure that near-term climate change impacts are moderated and that positive feedback responses—including many of those we discussed in this article—are limited.

SUMMARY POINTS

1. Atmospheric methane concentrations continue to increase.
2. Limitations of surface observation networks and inventory estimates make robust attribution of interannual variations difficult.
3. Global change, especially climate change, is a key determinant of global methane fluxes.
4. Climate change and carbon dioxide enrichment in the twenty-first century are likely to enhance methane emissions from large natural sources, such as wetlands and aquatic systems.
5. Climate change and carbon dioxide enrichment in the twenty-first century are likely to enhance methane emissions from some anthropogenic sources, such as hydroelectric energy generation.
6. Commercial exploitation of methane clathrate deposits has the potential to induce very large fugitive emissions.
7. There exists great potential for further, cost-effective (<\$100 per metric ton CO₂-e) climate change mitigation through reduced methane emissions in the energy, agriculture, and waste sectors.

FUTURE ISSUES

1. Enhanced surface observation networks, emissions inventories, and remote sensing products can improve methane source attribution and emissions management at national and subnational scales.
2. Integrating the response of global wetland methane fluxes to future climate change with that of local-scale human activities, such as wetland drainage and land use change, poses a significant modeling challenge.
3. Warming of the Arctic, coupled with technological developments in methane extraction, risks large-scale exploitation of methane clathrates and associated fugitive emissions to the atmosphere.
4. As methane emissions from the agriculture sector are dominated by ruminant livestock, future changes in population and dietary preference may radically alter national and global emissions from this source.

5. More assessment is required on the indirect impacts of global change on methane emissions via food loss and waste (e.g., increased food spoilage), ice retreat (e.g., increased accessibility to fossil methane deposits), and vegetation change (e.g., altered emergent vegetation in wetlands).

DISCLOSURE STATEMENT

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